	Chemistry				
	Year 12	Year 13			
A u t u m n	Elements of Life Atomic structure Mass Spectroscopy Nuclear fusion Wave and particle models of light Shells, subshells and orbitals Periodicity Covalent bonding and shapes of molecules A _r , M _r , percentage yield and balancing equations Bonding, structure, properties and precipitates Group1 and Group 2 chemistry	Polymers and life Amines, acyl chlorides and the formation of amides Hydrolysis of esters and amides Amino acids Optical stereoisomers Formation of peptide bonds Protein structure Interactions between drugs and receptor sites The structure and function of DNA and RNA Mass spectroscopy, NMR, combining all techniques			
A u t u m n 2	Elements of Life Reacting masses, acids and bases, neutralisation Concentration of solutions Developing Fuels Thermochemistry Enthalpy cycles and Hess' law Alkanes and shapes of molecules Bond enthalpies Catalysis Electrophilic addition, sigma and pi bonds Addition polymerisation Elements from the Sea The halogens Oxidation states and redox Electrolysis Dynamic equilibrium and K _c Redox titrations The risks and benefits of chlorine	The chemical industry Nitrogen chemistry (Redox in the soil and the laboratory) Equilibrium constant and the effect of changes Measuring rates of reactions Orders of reactions The Arrhenius equation Developing Metals Transition metals and their oxidation states Catalytic activity Colour in transition metal compounds and complexes Electrochemical cells			

	Developing Fuels	The chemical industry
S	Combustion of fuels, gas calculations, and the ideal gas equations	Half life method for finding order of reaction
	E/Z isomerism	Rate equations and mechanisms
р	Polluting gases	Industrial processes
r	Alternative fuels	madstrat processes
i		Developing Metals
n	Elements from the Sea	Rusting and methods of protection
g	Atom economy	Structure and properties of complexes
1	Hydrogen halides	
	Le chatelier's principle	
	The Ozone Story	Oceans
	Gas calculations	The dissolving process
	PPM	Lattice enthalpy
	Interactions of radiation with matter	Enthalpy change of solution
	Radical reactions	The greenhouse effect
S	Measuring rates of reactions	Acid-base chemistry
		pH calculations
p r	What's in a medicine	Buffer solutions
'	Reactions of alcohols	
· ·	The -OH group	Colour by Design
n g	Carboxylic acids	The origin of colour in organic molecules
g 2	Infrared spectroscopy	The delocalised model of benzene
-	Mass spectroscopy for compounds	Naming molecules containing benzene rings
		Electrophilic substitution reactions
		Making and modifying dye molecules
		dye-fibre interactions
		Fats and oils
		GLC
	The Ozone Story	Oceans
S	The effect of temperature on rate	Solubility products
u	Catalysis	Entropy
m	Haloalkanes	
m	Intermolecular bonding	Colour by Design
е	Nucleophilic substitution	Reactions of aldehydes and ketones
r		Planning an organic synthesis
1	What's in a medicine	
	Synthesis of aspirin	Preparation for examinations

	Principles of green chemistry	
S	Polymers and life	
u	Carboxylic acids, phenols, esters and condensation reactions	
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m		
е		
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