	Chemistry						
	Year 10 Separate	Year 10 Combined	Year 11 Separate	Year 11 Combined			
A u t m n 1	Structure and bonding Recap of Ionic, covalent and metallic bonding Giant ionic structures Structure of simple molecules Giant covalent structures	Structure and bonding Recap of Ionic, covalent and metallic bonding Giant ionic structures Structure of simple molecules	Rates and EquilibriaRate of reactionCollision theory and surface areaThe effect of temperatureThe effect of concentrationThe effect of catalystsReversible reactionsEnergy and reversible reactionsDynamic equilibriumAltering conditionsCrude oils and FuelsHydrocarbonsFractional distillation of oil	Rates and Equilibria Rate of reaction Collision theory and surface area The effect of temperature The effect of concentration The effect of catalysts Reversible reactions Energy and reversible reactions Dynamic equilibrium Altering conditions			
A u t n 2	Structure and bonding Fullerenes and graphene Giant metallic structures Nanoparticles and their application Chemical calculations Relative masses and moles Equations and calculations Percentage yield Atom economy	Structure and bonding Giant covalent structures Fullerenes and graphene Giant metallic structures Chemical calculations Relative masses and moles Equations and calculations	Crude oils and Fuels Burning hydrocarbon fuels Cracking hydrocarbons Organic reactions Reactions of alkenes Structures of alcohols, carboxylic acids and esters Reactions and uses of alcohol Carboxylic acids and esters Polymers Addition polymerisation Condensation polymerisation Natural polymers DNA	Crude oils and Fuels Hydrocarbons Fractional distillation of oilBurning hydrocarbon fuels Cracking hydrocarbons			

	Chemical calculations	Chemical calculations	Chemical Analysis	Chemical Analysis
	Expressing concentrations	Expressing concentrations	Chemical analysis	Chemical analysis
	Titrations		Analysing chromatograms	Analysing chromatograms
	Gas volumes		Testing for gases	
		Chemical changes	Testing for positive and negative ions	
	Chemical changes	The reactivity series	Instrumental analysis	The Earth's atmosphere
	The reactivity series	Displacement reactions		History of the atmosphere
S	Displacement reactions	Extracting metals	The Earth's atmosphere	The evolving atmosphere
р	Extracting metals		History of the atmosphere	Greenhouse gases
r			The evolving atmosphere	Climate change
i			Greenhouse gases	Atmospheric pollutants
n			Climate change	
g			Atmospheric pollutants	
1				
			The Earth's resources	
			Finite and renewable resources	
			Water safe to drink	
			Treating waste water	
			Extracting metals from ores	
			Life cycle assessments	
S	Chemical changes	Chemical changes	Using resources	The Earth's resources
р	Salts from metals, metal oxides, metal	Salts from metals, metal oxides, metal	Rusting	Finite and renewable resources
r	carbonates	carbonates	Useful alloys	Water safe to drink
i	Neutralisation and the pH scales	Neutralisation and the pH scales	The properties of polymers	Treating waste water
n	Strong and weak acids	Strong and weak acids	Glass, ceramics and composites	Extracting metals from ores
g			The Haber process	Life cycle assessments
2			Making fertilisers	
S	Electrolysis	Electrolysis	Revision	Revision
u	Basics of electrolysis	Basics of electrolysis		
m	Extraction of aluminium	Extraction of aluminium		
m	Changes at the electrode	Changes at the electrode		
е	Electrolysis of aqueous solutions	Electrolysis of aqueous solutions		
r				
1				

S	Energy changes	Energy changes
u	Exothermic and endothermic energy	Exothermic and endothermic energy
m	changes	changes
m	Reaction profiles	Reaction profiles
e	Bond energy calculations	Bond energy calculations
r	Chemical cells and batteries	
2	Fuel cells	